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Titration Gizmo : ExploreLearning. Measure the quantity of a known solution needed to neutralize an acid or base of unknown concentration. Use this information to calculate the unknown concentration. A variety of indicators can be used to show the pH of the solution.

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Titration - for Gr 11 Chemistry, University Prep . I adapted the student worksheets to meet the expectations of the Ontario Curriculum. I removed the titrat... (more) ion curves and set it up as a "discovery" exploration. Students practice a titration before completing the real thing and incorporate stoichiometry for neutralization reactions.

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Student Exploration: Titration neutralize, pH, strong acid, strong base, titrant, titration, titration curve, weak acid, weak base Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

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Student Exploration Titration Gizmo Answer Key Activity B

In the Titration Gizmo™, you will use indicators to show how acids are neutralized by bases, and vice versa. To begin, check that 1.00 M NaOH is selected for the Burette, Mystery HBr is selected for the Flask, and Bromthymol blue is selected for the Indicator. 1. Look at the flask.

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